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Homogeneous Catalysis

Distannylation of Strained Carbon-Carbon Triple Bonds Catalyzed by a Palladium Complex**

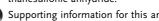
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Transformations of arynes catalyzed by a transition-metal complex, although less common than stoichiometric reactions,[1] have recently received considerable attention as a novel route for the synthesis of substituted arenes that are not accessible by conventional methods. Irrespective of the catalyst and reactant used, efforts have thus far been devoted mainly to developing cyclization reactions of arynes. The pioneering work on these transformations-palladium-catalyzed trimerization of arynes^[2] and cocyclization of arynes

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with alkynes^[3]—was reported by Peña et al. Catalytic reactions of arynes with alkynes and/or allyl chlorides in the presence of a palladium complex have also been reported by Yamamoto and co-workers, [4] while Murai and co-workers have developed cobalt-, rhodium-, and palladium-catalyzed carbonylative cycloadditions of arynes. [5] Catalytic addition reactions of a σ bond [6] to arynes have been limited to carbostannylation [7a] and disilylation, [7b] as reported previously. Herein we report that a tin–tin bond of a distannane readily adds to strained carbon–carbon unsaturated compounds, such as arynes or cyclohexynes, in the presence of a palladium complex [Eq. (1)]. This transition-metal-catalyzed element–element σ -bond addition reaction to cyclohexynes has no precedents, to the best of our knowledge.

$$R \xrightarrow{+ \frac{Sn}{Sn}} + \frac{Sn}{Sn} \xrightarrow{[Pd]} R \xrightarrow{Sn} (1)$$

We first examined the reaction of hexabutyldistannane (1a) and benzyne, prepared in situ from 2-(trimethylsilyl)phenyl triflate (2a),[8] and a fluoride ion (KF/[18]crown-6) in the presence of a palladium/1,1,3,3-tetramethylbutyl isocyanide (tert-octyl isocyanide, tOcNC) complex, [9] and found that the tin-tin bond of 1a smoothly adds to benzyne to afford 1,2bis(tributylstannyl)benzene (3aa) in 73% yield (entry 1, Table 1).^[10] The choice of catalyst and source of the fluoride ion are crucial: the reaction did not proceed in the presence of either [Pd(dba)₂] (dba = trans,trans-dibenzylideneacetone) or [Pd(PPh₃)₄], and the desired product was also not formed when CsF in MeCN^[11] was used as the source of the fluoride ion, even in the presence of the same palladium-isocyanide complex. The distannylation of various arynes was then investigated under the optimized reaction conditions. As in the case of benzyne itself, the 4-substituted benzynes formed in situ from precursors 2b-2d react efficiently with 1a to

provide the corresponding distannylation products 3ab-3ad, respectively, in moderate to high yields (entries 2-4), whereas the reaction of 4-fluorobenzyne (formed from 2e) gave only a low yield of the desired product (entry 5). The distannylation reaction was also applied to 4,5-disubstituted benzynes (formed from 2 f or 2 g) or sterically more congested arynes (formed from 2h-2j) to afford good to high yields of products **3af-3aj**, respectively (entries 6–10). In addition to **1a**, hexamethyldistannane (1b) could also be used as the tincontaining starting material, and in this case the less-reactive 1,2-naphthalyne (from 2k), 3-phenylbenzyne (from 2l), or 9,10-phenanthryne (from 2m) were found to insert into the tin-tin bond (entries 11-14, respectively); [12] in most cases a compound resulting from cyclotrimerization of the aryne (triphenylene) was formed as a by-product. The reaction with 1b also produced a trace amount of 2,2'-distannylbiaryl as well; for example, triphenylene and 2,2'-bis(trimethylstannyl)biphenyl were obtained in 9% and 3% yields in the reaction summarized in entry 11.

The reaction of an aryne containing two triple bonds is noteworthy: treatment of the bisaryne precursor **2n** with **1a** gave bis[3,4-bis(tributylstannyl)phenyl] ether **3an** in which four carbon–tin bonds are formed in one step [Eq. (2); Tf = trifluoromethanesulfonyl, TMS = trimethylsilyl].

Table 1: Distannylation of arynes.

$$\begin{array}{c} \text{KF (3 equiv)} \\ \text{[18]crown-6 (3 equiv)} \\ \text{Pd(OAc)}_2 \text{ (2 mol\%)} \\ \text{Pd(OAc)}_2 \text{ (2 mol\%)} \\ \text{HocNC (30 mol\%)} \\ \text{THF, 20 °C} \\ \text{SnR}_3 \\ \text{SnR}_3$$

Entry	Distannane	R' in 2	Precursor	<i>t</i> [h]	Yield [%] ^[a]	Product
1	1a (R = Bu)	Н	2 a	3	73	3 aa
2	` , ,	4-Me	2 b	3	71	3 ab
3		4-MeO	2 c	7	55	3 ac
4		4-Ph	2 d	4	40	3 ad
5		4-F	2 e	6.5	26	3 ae
6		4,5-Me ₂	2 f	3	73	3 af
7		4,5(CH ₂) ₃ -	2 g	9	64	3 ag
8		6-Me	2h	3.5	55	3 ah
9		3-MeO	2i	6	59	3 ai
10		3,6-(MeO) ₂	2j	5	63	3 aj
11	1b (R = Me)	Η ` /-	2 a	4	56	3 ba
12	,	3,4(CH) ₄ -	2 k	4.5	61	3 bk
13		6-Ph	21	5.5	39	3 bl
14		dibenzo[c,e] ^[b]	2 m	5	27	3 bm

[a] Yield of isolated product based on distannane. [b] 9,10-Phenanthryne.

The distannylation reaction was also found to be suitable for cyclohexynes (Scheme 1). When in situ generated cyclohexyne (formed from 20) was treated with 1a under the above-mentioned reaction conditions the insertion product 1,2-bis(tributylstannyl)cyclohexene (3a0) was obtained in 70% yield. Similarly, the reaction of substituted cyclohexynes (formed from 2p or 2q) with 1b also took place successfully to provide the distannylation products (3bp or 3bq) in 60 or 90% yield, respectively.

Since an aryl (or alkenyl) triflate readily undergoes a cross-coupling reaction with a distannane in the presence of a palladium catalyst, [13] the distannylation might proceed by a pathway that does not

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Scheme 1. Distannylation of cyclohexynes: a) KF (3 equiv), [18]crown-6 (3 equiv), Pd (OAc)₂ (2 mol%), tOcNC (30 mol%), THF, 20°C.

involve an aryne (cyclohexyne) intermediate: cross-coupling of **2** at a C-OTf moiety with **1** followed by fluoride-ion-induced silicon-tin exchange between the resulting 2-(trial-kylstannyl)arylsilane (2-(trialkylstannyl)cyclohexenylsilane) and R₃SnOTf.^[14] However, this pathway can be ruled out since the reaction of 4-(trimethylsilyl)phenyl triflate (**2r**) with **1a** did not give the respective distannylation product, thus confirming the intermediacy of an aryne (cyclohexyne) in the distannylation process [Eq. (3)].

We assumed that this reaction is initiated by oxidative addition of an Sn–Sn bond of a distannane to a Pd⁰ complex^[15] and thus we examined the stoichiometric reaction shown in Scheme 2. Thus, treatment of the Pd⁰-isocyanide complex^[16]

Scheme 2. Elucidation of the mechanism by stoichiometric reactions: a) 2a (1.5 equiv), KF (3 equiv), [18]crown-6 (3 equiv), THF, 20°C, 1 h.

with an equimolar amount of **1a** led to the quantitative formation of the oxidative-addition product **4a**, [17,18] which exhibits high reactivity for the insertion of benzyne to afford **3aa** in 74% yield. These results prompted us to propose cycle A depicted in Scheme 3. First, complex **4** is produced by oxidative addition of **1** to the Pd⁰ complex. Subsequent

Scheme 3. Catalytic cycles for the distannylation.

insertion of the triple bond of the aryne or cyclohexyne into an Sn-Pd bond of **4** to form palladium complex **5**, followed by reductive elimination, affords the product and regenerates the catalyst. However, formation of a triphenylene and distannylbiaryl as by-products may suggest that another catalytic cycle is also operative. Thus, the Pd⁰ complex initially undergoes oxidative cyclization with the aryne to generate palladacycle **6**,^[19] which yields the product after reaction with **1** (cycle B).^[20,21] The by-products could also arise from **6**: the interaction of **6** with a second aryne instead of **1** gives a five-membered palladacycle **7**, which reacts further with a third aryne^[22] or **1**^[23] to afford a triphenylene or distannylbiaryl, respectively.

In conclusion, distannylation of arynes or cyclohexynes has been accomplished in the presence of a catalytic amount of a palladium–*tert*-octyl isocyanide complex, and diverse 1,2-distannylarenes or 1,2-distannylcyclohexenes, which could have high synthetic utility as bis(anion) equivalents, have been readily synthesized. Moreover, the catalytic pathway, which includes oxidative addition of a distannane to a palladium(0) complex, has been confirmed by the stoichiometric reaction. Further studies on synthetic applications to other strained cycloalkynes and σ bonds are in progress.

Experimental Section

3aa: Compounds **1a** (0.13 g, 0.22 mmol) and **2a** (0.098 g, 0.33 mmol) were added to a solution (1.0 mL) of 1,1,3,3-tetramethylbutyl isocyanide (9.2 mg, 0.066 mmol), Pd(OAc)₂ (1.0 mg, 4.4 μmol), KF (0.038 g, 0.66 mmol) and [18]crown-6 (0.17 g, 0.66 mmol) in THF, and the resulting mixture was stirred at 20 °C for 3 h. The mixture was then diluted with ethyl acetate, filtered through celite, and concentrated. Column chromatography on alumina (hexane as eluent, activity IV) followed by gel-permeation chromatography (benzene as eluent) gave **3aa** as a colorless oil (0.11 g, 73% yield); ¹H NMR (400 MHz, CDCl₃): δ = 0.88 (t, J = 7.1 Hz, 18 H), 1.01–1.07 (m, 12 H), 1.27–1.53 (m, 24 H), 7.19–7.22 (m, 2 H), 7.42–7.55 ppm (m, 2 H); ¹³C NMR (99.5 MHz, CDCl₃): δ = 10.9, 13.6, 27.5, 29.2, 127.1, 137.5, 151.9 ppm; ¹¹⁹Sn NMR (147.5 MHz, CDCl₃): δ = −43.2 ppm. Elemental analysis calcd for C₃₀H₅₈Sn₂: C 54.91, H 8.91; found: C 55.18, H 8.97

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